

Human acceleration of the nitrogen cycle at regional to global scales

Bob Howarth Ecology & Evolutionary Biology

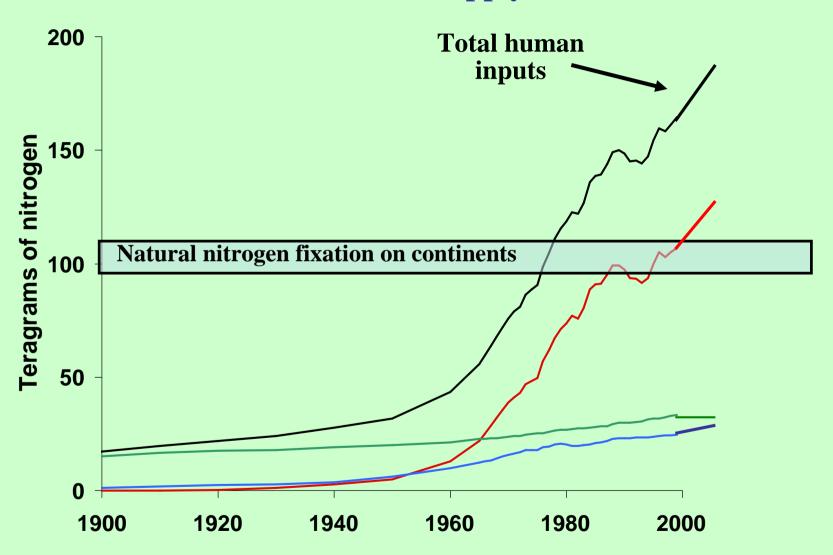
Computational Sustainability

10 March 2010



The planet has a fever The green line represents the upper safe limit of each of the earth's 10 life-sustaining biophysical systems. Where "mercury" has risen above that line, humanity has already transgressed the boundary. risking potentially irreversible "tipping points." tipping point CLEBATE CHRISTE CHRISTE THE CHCLE CH Source: Nature, "Planetary Boundaries: A Safe Operating Space for Humanity." 24 Sept. 2009

Global trends in nitrogen use per year – humans have doubled the natural supply



(modified from Howarth et al. 2005)

What's the problem?

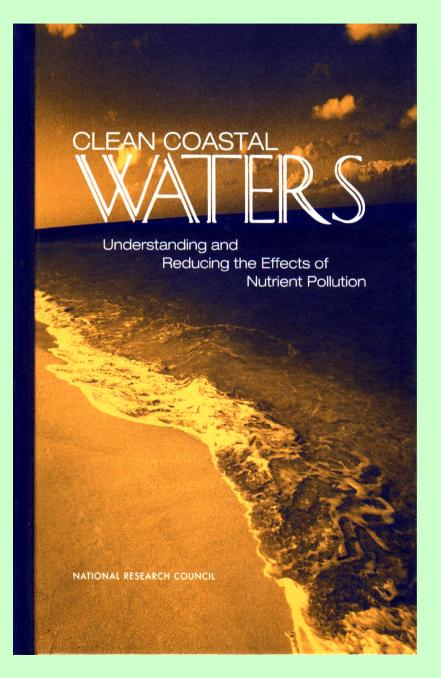
Nitrogen is a nutrient...

Nutrients are "things that nourish."

Too much of a good thing!







NRC 2000:

Nitrogen is now the largest pollution problem in the coastal waters of the United States.

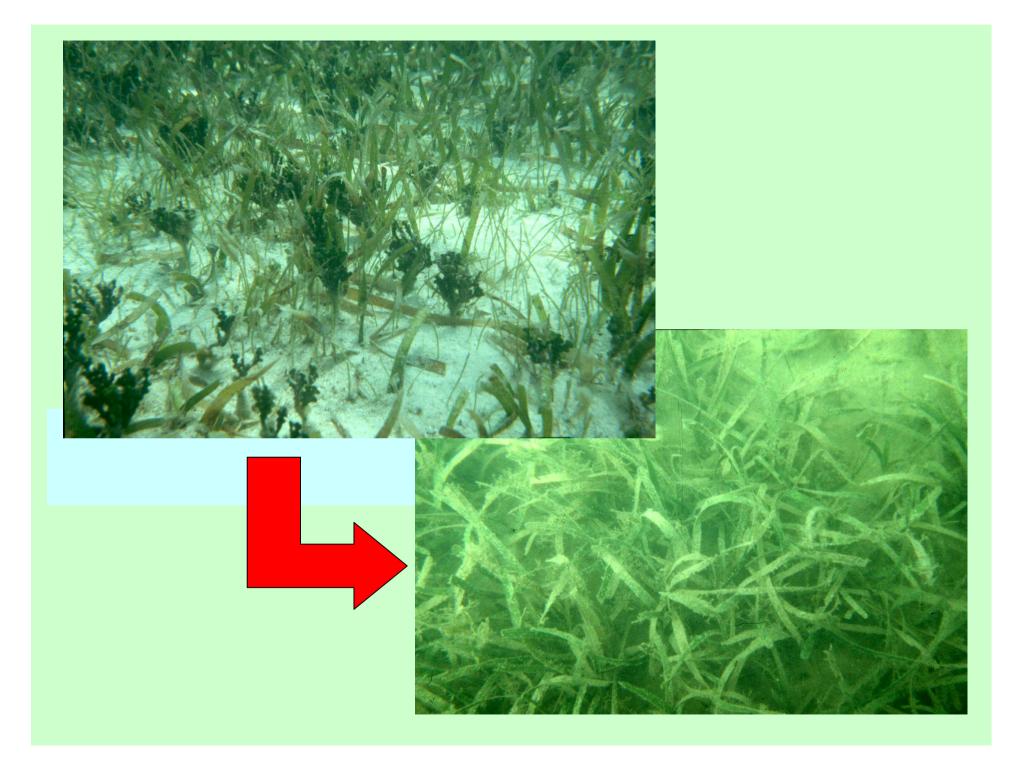
Two thirds of coastal rivers and bays are moderately to severely degraded from nitrogen pollution.

"Dead Zones" (low-oxygen waters) from Excess Nitrogen Pollution



United Nations Environmental Programme (2008)







Townsend & Howarth (2010) Scientific American

Other environmental consequences of accelerated N cycle

- Acid rain: damage to waters and soils
- Damage to growth of forests and agriculture, due to ozone pollution
- Over-stimulation of forest growth, leading to winter dieback
- Loss of biodiversity in forests, wetlands, etc.

Other environmental consequences of accelerated N cycle

	Concentrations ^b ar	nd their changes ^c	Radiative I	liative Forcing ^d		
Species ^a	2005	Change since 1998	2005 (W m ⁻²)	Change since 1998 (%)		
CO2	379 ± 0.65 ppm	+13 ppm	1.66	+13		
CH ₄	1,774 ± 1.8 ppb	+11 ppb	0.48	-		
N ₂ O	319 ± 0.12 ppb	+5 ppb	0.16	+11		

Nitrous oxide gas (N_2O) – no laughing matter!

- greenhouse gas -- much lower concentration that CO₂, but 280-fold more powerful! Responsible for about 1/10th of warming currently, compared to CO₂
- creates ozone holes in stratosphere, letting high UV radiation hit surface of earth
- stays in the atmosphere for 120 years.

Nitrogen and Human Health

Direct pollution effects:

- Heart disease and lung diseases (from gases and particles in air)
- Cancer (nitrogen in drinking water)

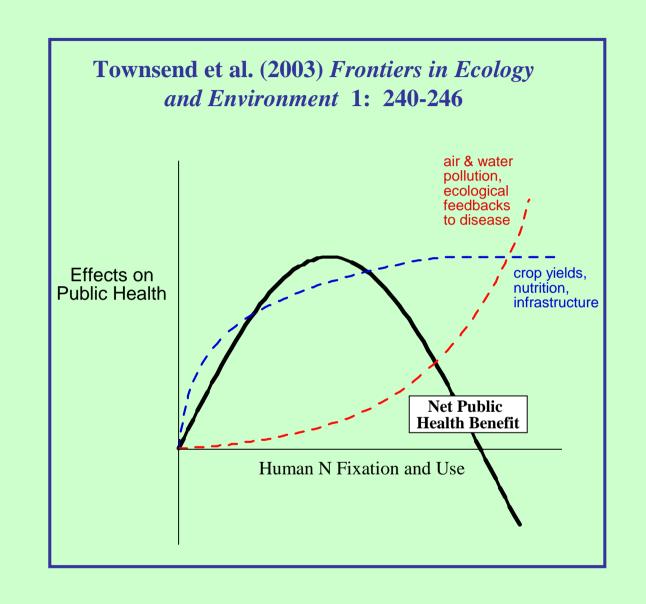
Indirect pollution effects (from fertilization!):

- Pollen and allergies
- •Animals that carry diseases (mosquitoes, snails) such as malaria and schistosomiasis

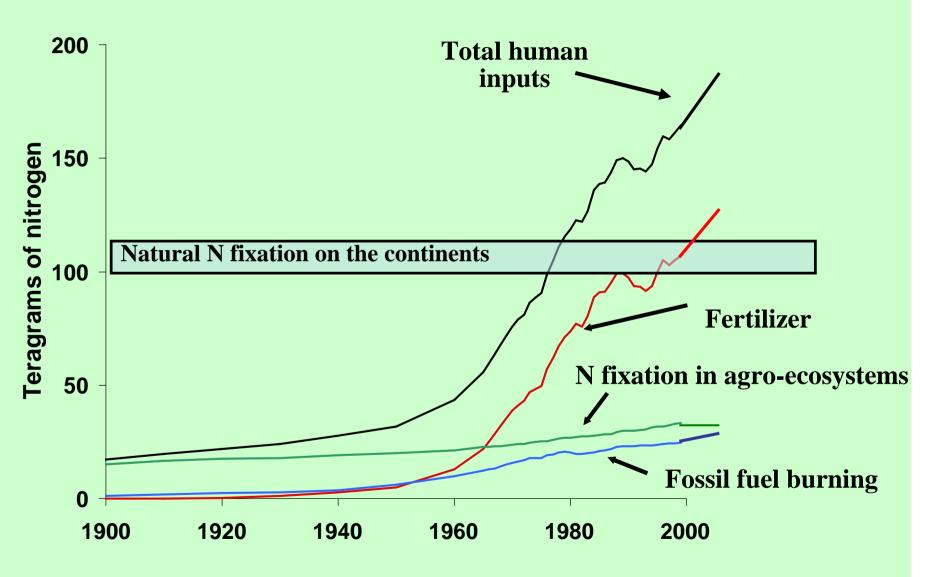








Global trends in nitrogen use per year

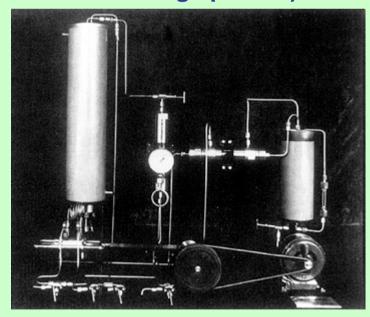


(modified from Howarth et al. 2005)

Fritz Haber discovered a way to make synthetic nitrogen fertilizer from atmospheric N_2 gas in 1909.

Today, this literally feeds the world!

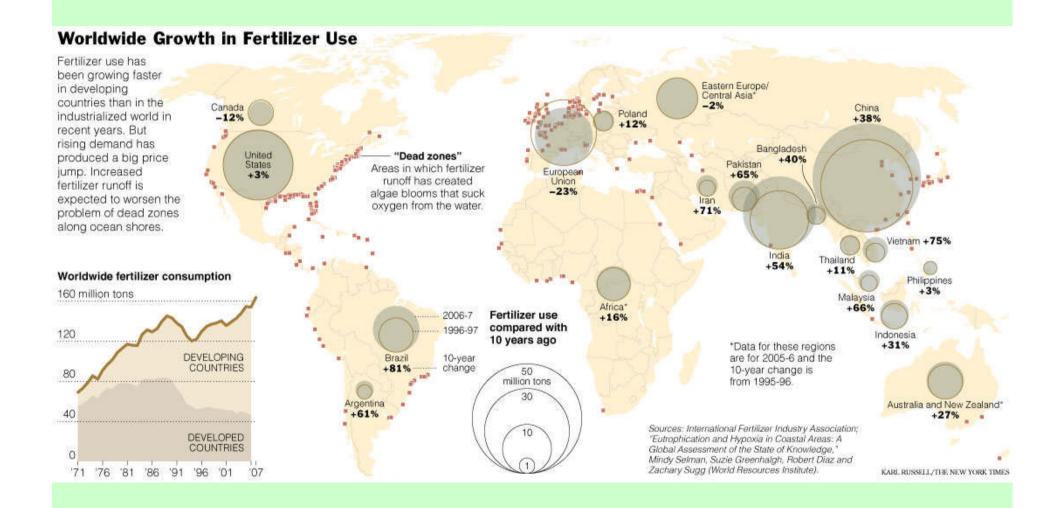
80% of the nitrogen that supports food production comes from this industrial process (and 80% of the nitrogen in your proteins, if you eat like an average person).





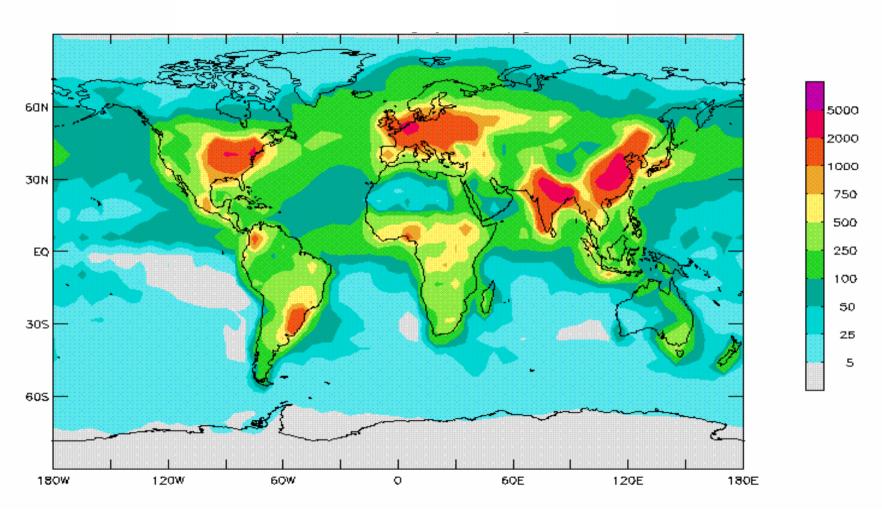
Burning of fossil fuels contributes..... and is a bigger problem that agriculture in some parts of the northeastern US. But in most of the world, agriculture is the biggest source of nitrogen pollution.

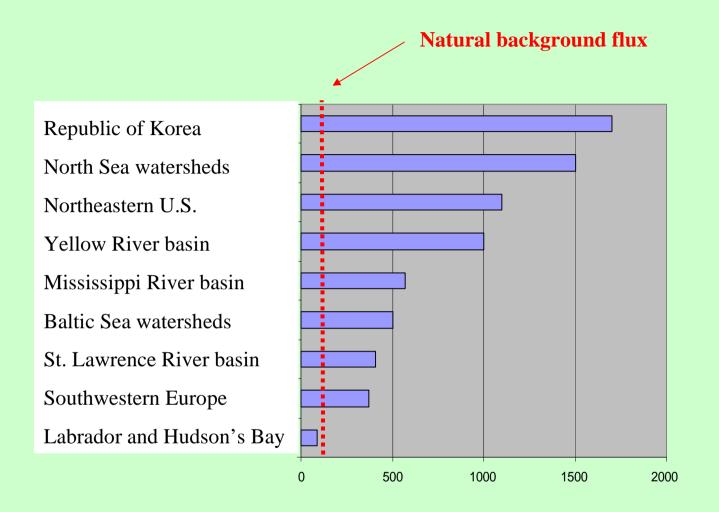




New York Times (4/30/08)

Total nitrogen deposition as of 1993 (kg km⁻¹yr⁻¹)





Flux of nitrogen from the landscape to coastal oceans in rivers for contrasting regions of the world in the temperate zone (kg per km² of watershed area per year; from Howarth et al. 2005).

For particular regions and watersheds, where does the nitrogen come?

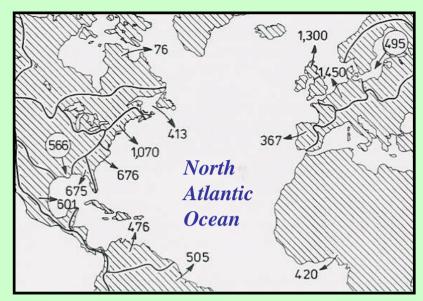
Only models provide an answer.....

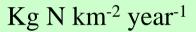
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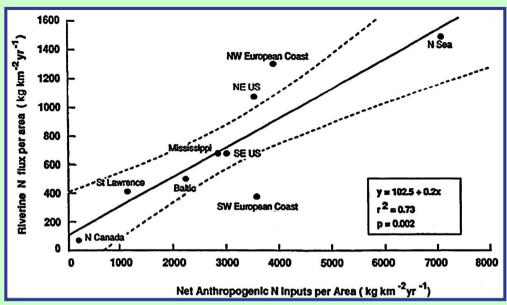
Only models provide an answer.....

Many available models, and we use a few different ones, of varying complexity. Today, I will focus on the simplest: large spatial scales, multiple-year average input data.

(This forms the basis for some of the more complicated models, which run at daily time scales and demand more types of data input).



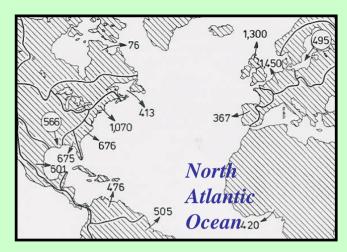


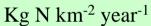


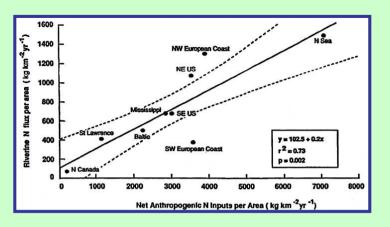
	NO _y deposition	Fertilizer	N fixation by crops	Net import (+) or expor in foods	Total t
North Canada rivers	70	160	30	-50	210
St. Lawrence basin	610	330	260	-30	1170
NE coast of US	1200	600	750	1000	3550
SE coast of US	1020	1170	370	450	3010
Eastern Gulf of Mexico	760	1260	250	580	2850
Mississippi River basin	620	1840	1060	-1300	2220
Baltic Sea drainages	480	1730	30	20	2220
North Sea drainages	1090	5960	5	-5	7050
NW European coast	1090	2870	50	-320	3700
SW Eutropean coast	460	3370	15	-65	3780

NANI (net anthropogenic N inputs)

(Howarth et al. 1996)





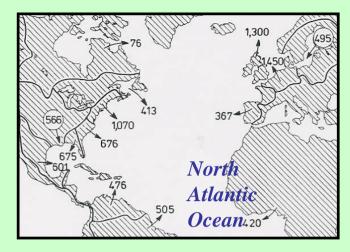


(Howarth et al. 1996)

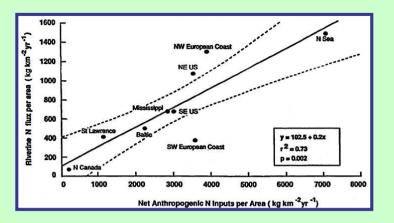
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NANI:

- only 4 inputs
- deposition of NH₃ and NH₄+, wastewater inputs, and manure are not included (these are considered recycling within region)



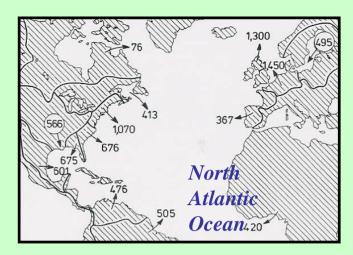
Kg N km⁻² year⁻¹



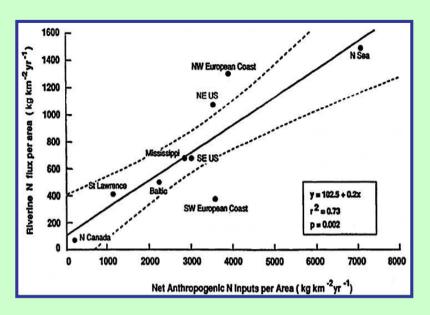
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These inputs are easier to estimate at larger spatial scales



Kg N km⁻² year⁻¹



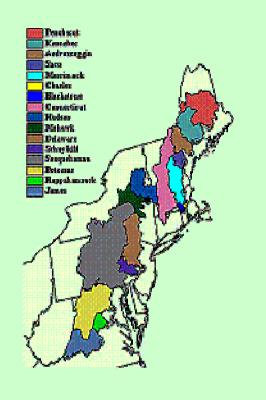
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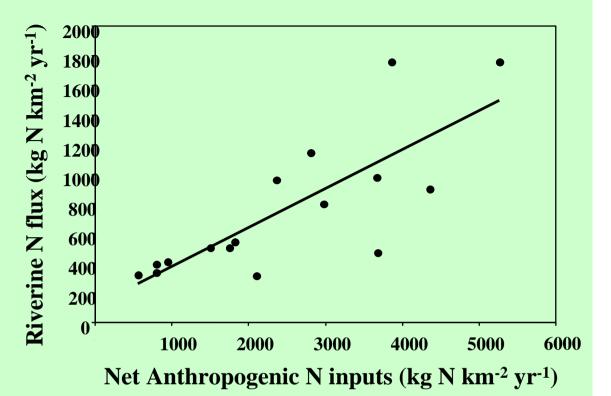


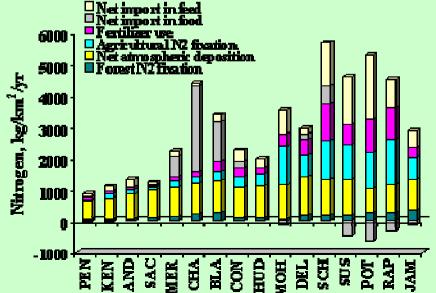
NANI compared to riverine TN flux to coast over periods of ~ 6 years, to average out year-to-year variability.

• The simple NANI model compares extremely well with more complicated models for estimating N export from large watersheds and regions (Alexander et al. 2002; Boyer & Howarth 2008).

• We do even better by including a little more information (climate or hydrology).

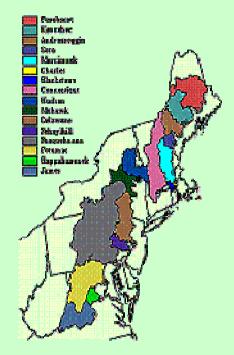




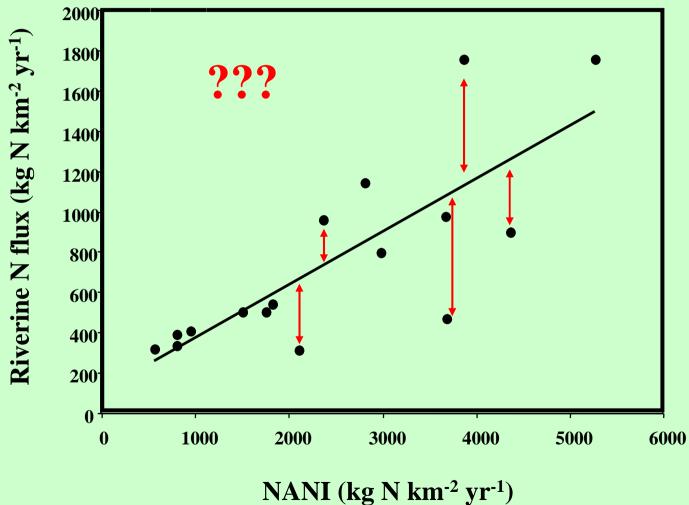


Application to individual watersheds in northeastern US

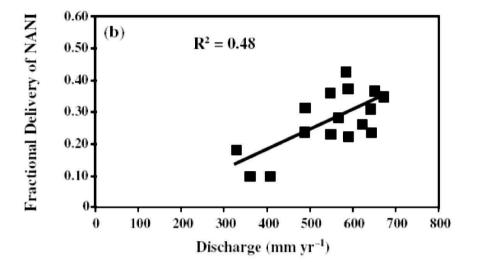
(Boyer et al. 2002; Howarth et al. 2006)

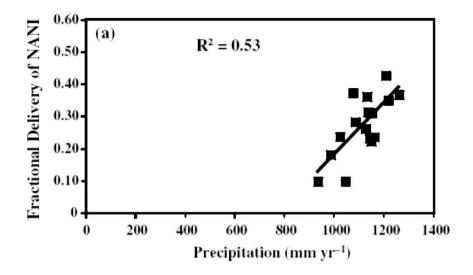


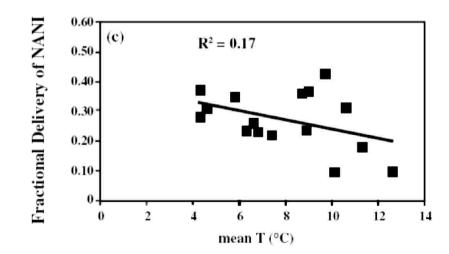
Is scatter attributable to climate differences?



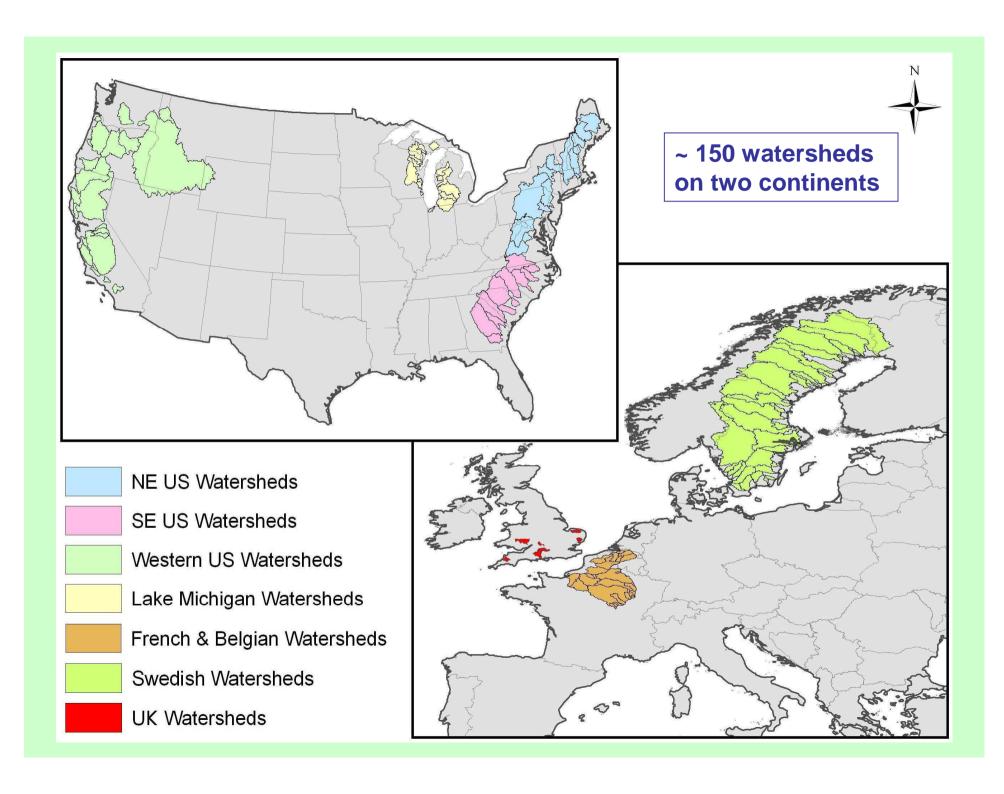
Fractional delivery = (river flux) / (N inputs)

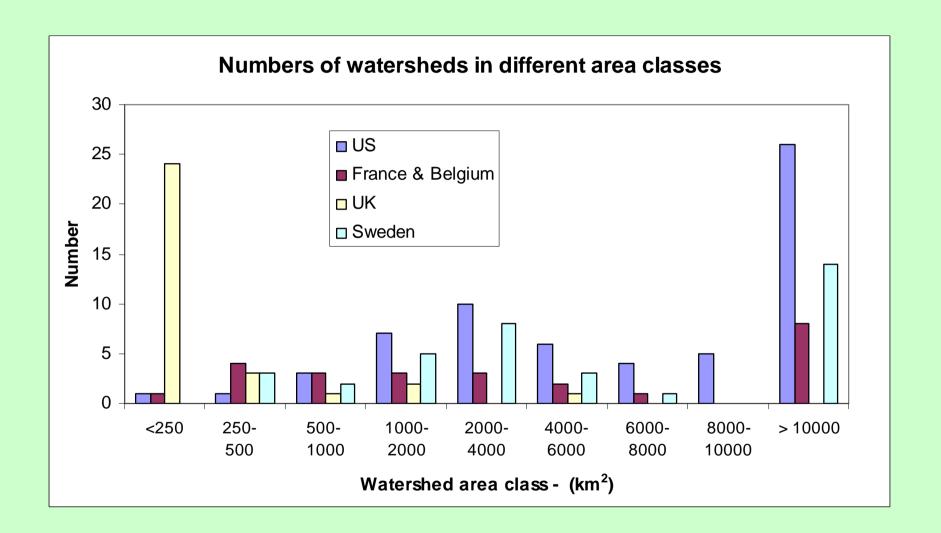


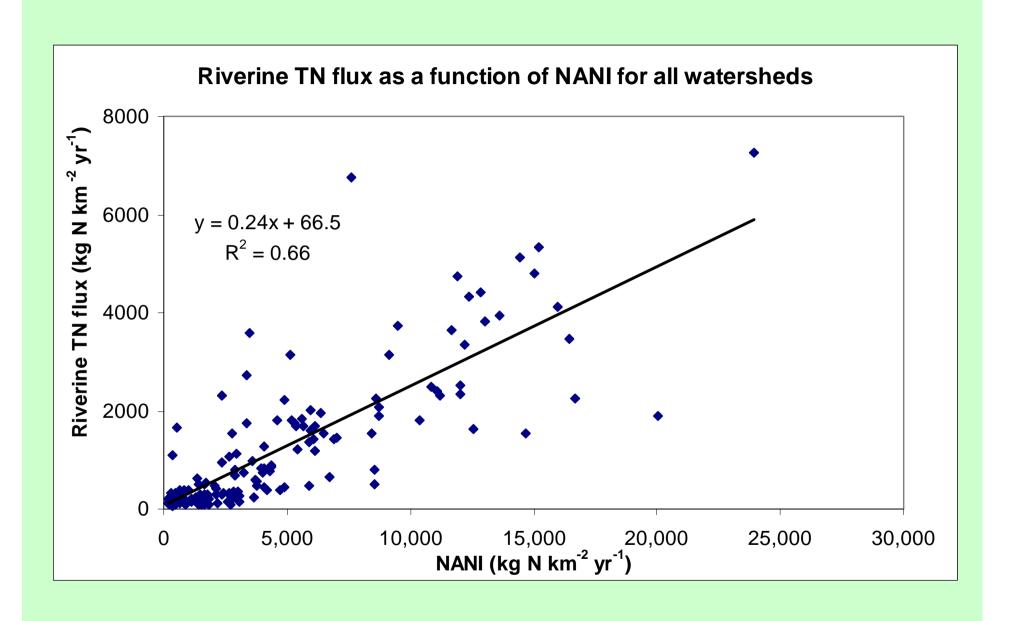




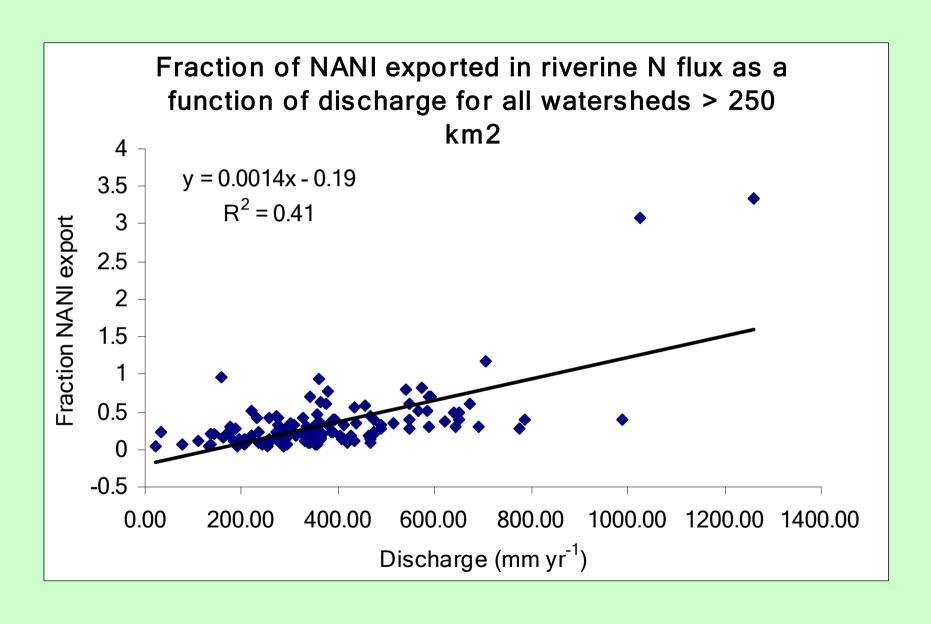
Howarth et al., 2006







Fraction of NANI Export as a Function of Discharge



Models with intercepts set to zero:

#2) TN flux =
$$b^*$$
 NANI + c^* NANI*Q + d^* Temp (adj.R2 =0.86)

	NANI*Q	<u>NANI</u>	<u>Temp</u>
Coefficients	0.00033	0.13	-4.77
P-value	6.5E-10	4.9E-07	0.48

#4) TN flux =
$$b^*$$
 NANI + c^* NANI*Temp+ d^* Temp (adj.R2 =0.82)

	<u>NANI</u>	<u>Temp</u>	NANI*Temp
Coefficients	0.355	-3.16	-0.010
P-value	6.95E-10	0.71	0.064

#6) TN flux =
$$b^*$$
 NANI + c^* NANI*Q (R2 =0.86)

	<u>NANI*Q</u>	<u>NANI</u>
Coefficients	0.00034	0.118
P-value	3.08E-10	4.95E-08

#8) TN flux =
$$b^*$$
 NANI + c^* NANI*Q+ d^* Q (R2 =0.86)

	NANI*Q	<u>NANI</u>	<u>Q</u>
Coefficients	0.00032	0.116	0.13
P-value	5.6E-09	9.1E-08	0.36

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NANI*Q NANI 0.00034 0.118

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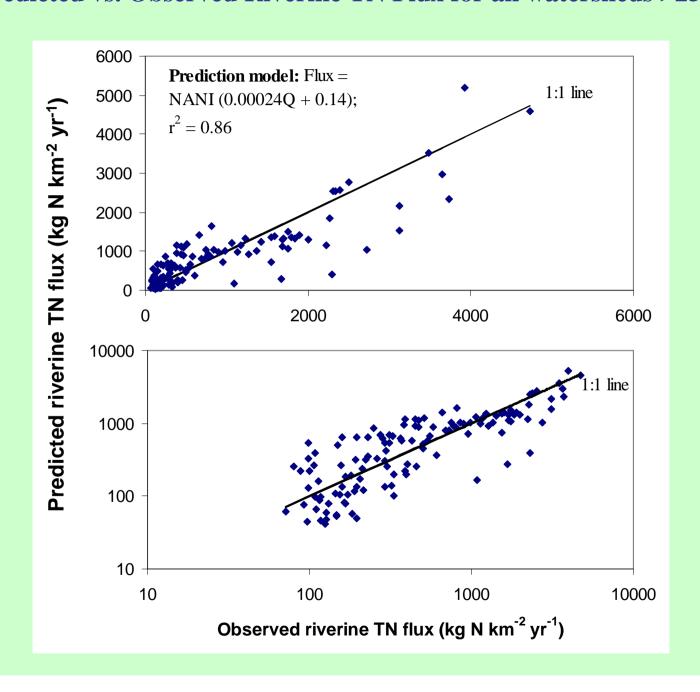
Coefficients

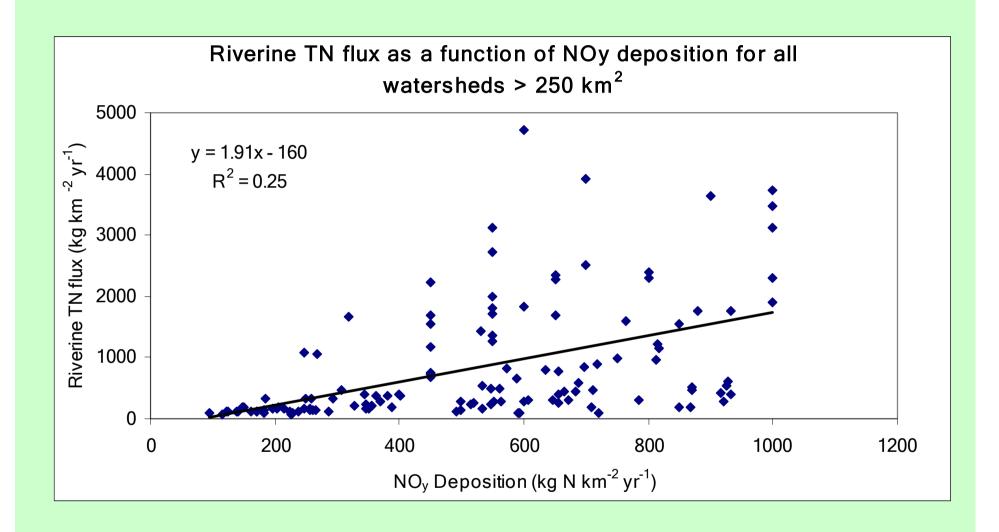
High R2 and all terms significant

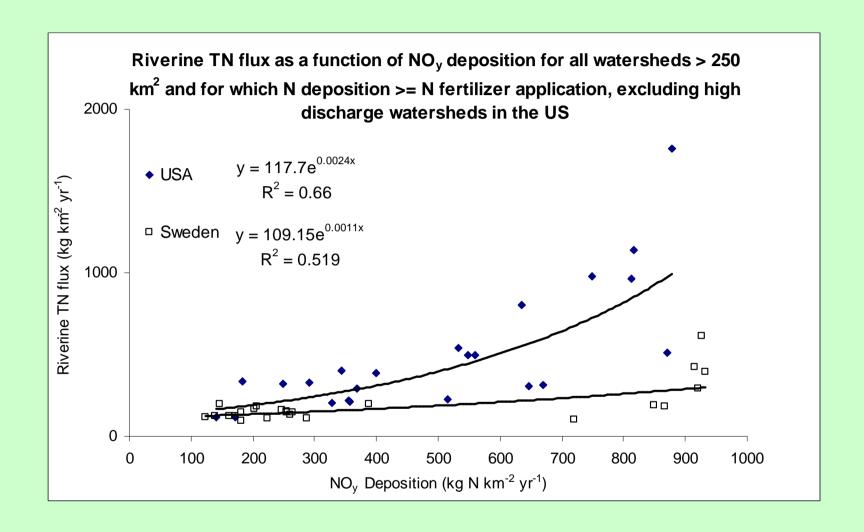
#8) TN flux =
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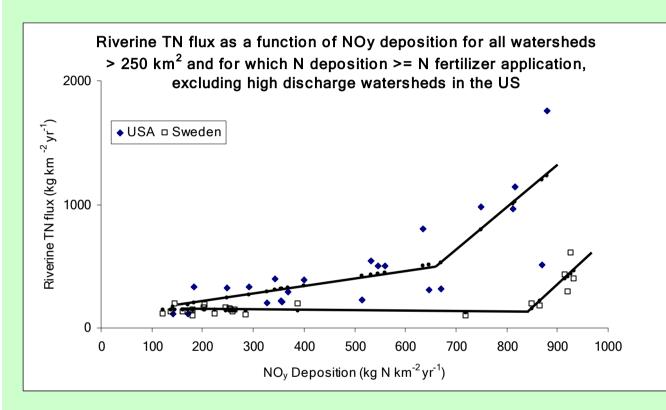
	NANI*Q	<u>NANI</u>	<u>Q</u>
Coefficients	0.00032	0.116	0.13
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Predicted vs. Observed Riverine TN Flux for all watersheds >250 km²









The piecewise linear functions were fit by minimizing the sum of the squared differences between the data and the model.

The best fit:

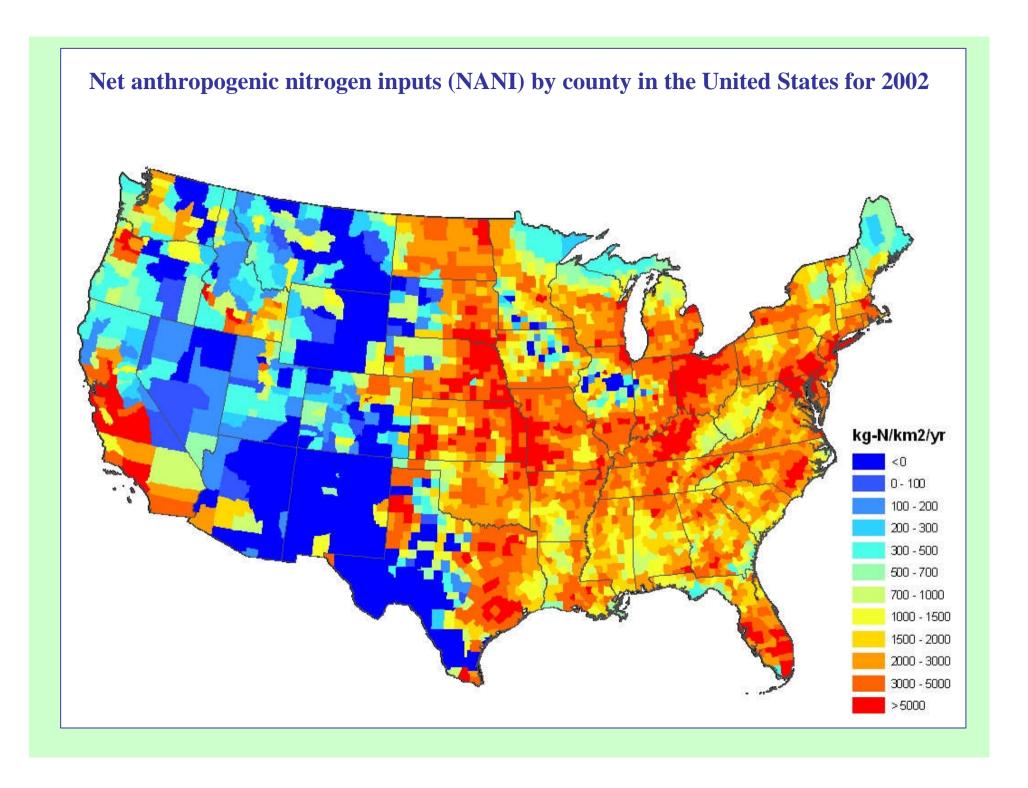
	US	Sweden
r2	0.67	0.80
а	76.27	149.98
b	0.67	-0.04
С	3.39	3.64
X _t	670.73	838.91

$$y = a + b x , x < x_t$$

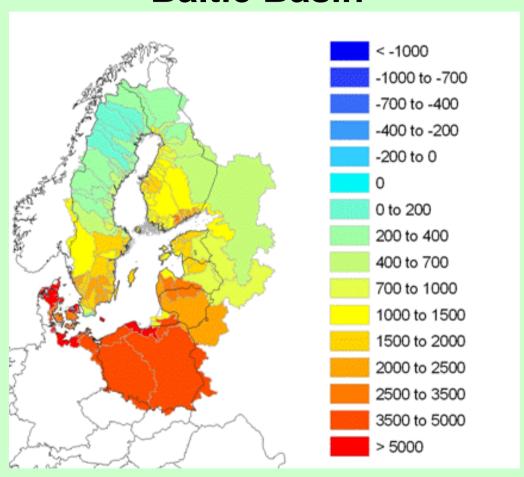
$$y = a + b x_t + c (x-x_t), x>= x_t$$

parameters:

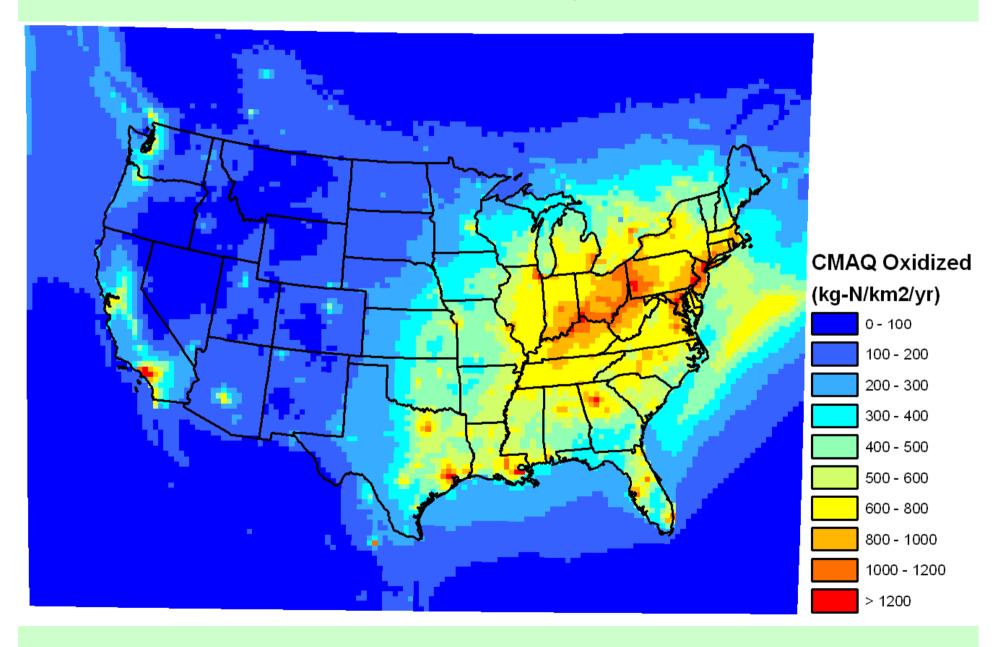
- a y intercept
- b slope of the first piecewise segment, $x < x_t$
- c slope of the second piecewise segment, x >= x,
- x_t threshold value of x at which the slope changes



NANI (kg/km2/yr) Baltic Basin

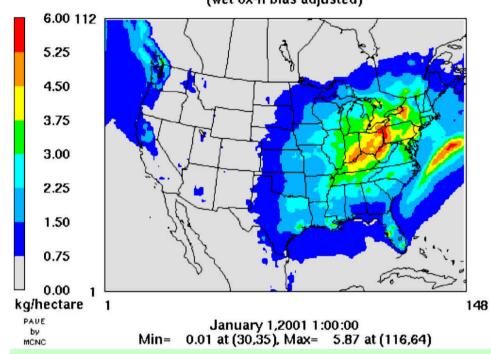


Atmospheric Deposition (wet + dry) of Oxidized Nitrogen estimated from CMAQ model



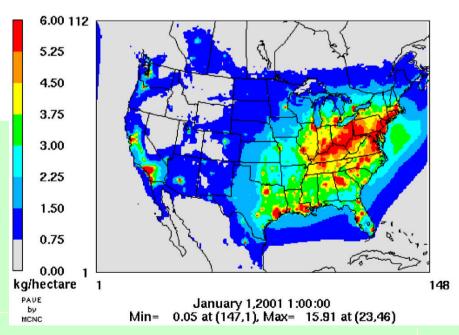
WET OXIDIZED NITROGEN

2001 BASE (J4f) - ANNUAL (wet ox-n bias adjusted)



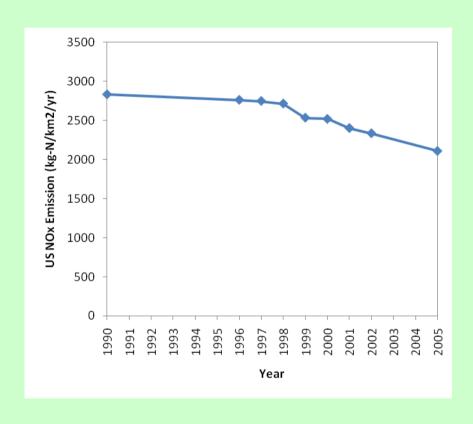
DRY OXIDIZED NITROGEN

2001 BASE (J4f) - ANNUAL

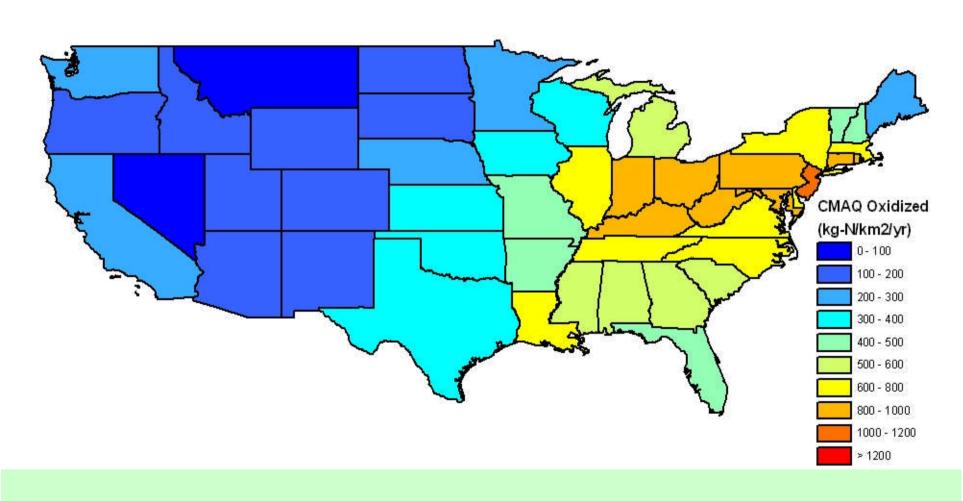


Beta version of CMAQ model (courtesy of Robbin Dennis, NOAA/EPA)

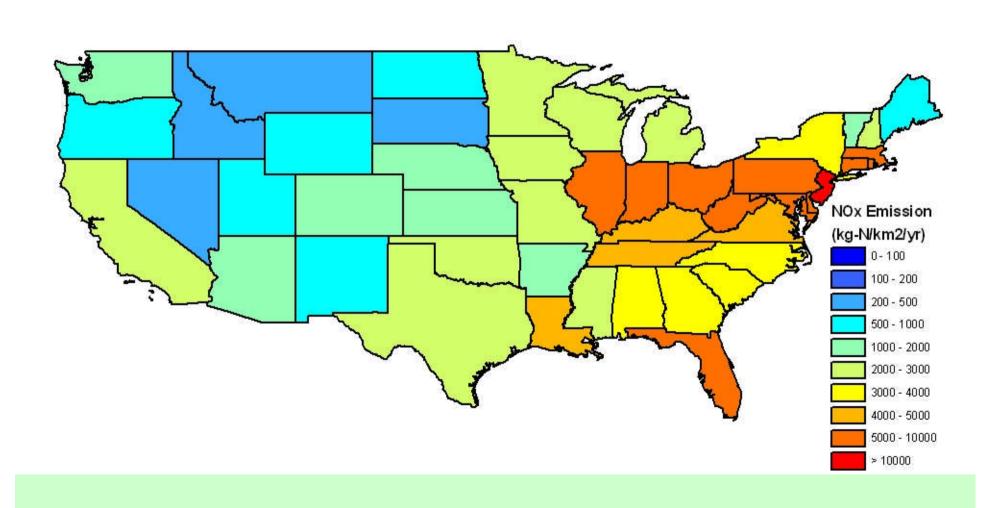
US oxidized N emissions



2002 estimated state level oxidized N deposition (EPA CMAQ model estimates)



2002 state level oxidized N emissions



Summarizing complex models with relatively simple statistical relationships

Analyze relationship between state-level N emission and deposition components with ordinary least-squares regressions:

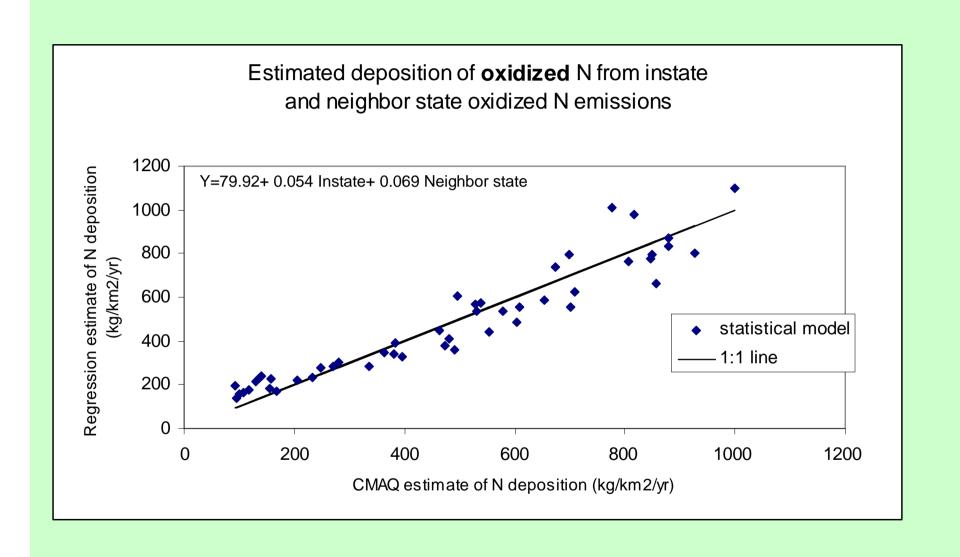
Using state-level data:

Oxidized N: Ndepox = $\frac{a}{b}$ *NOx(instate)+ $\frac{c}{c}$ NOx(neighbor states)

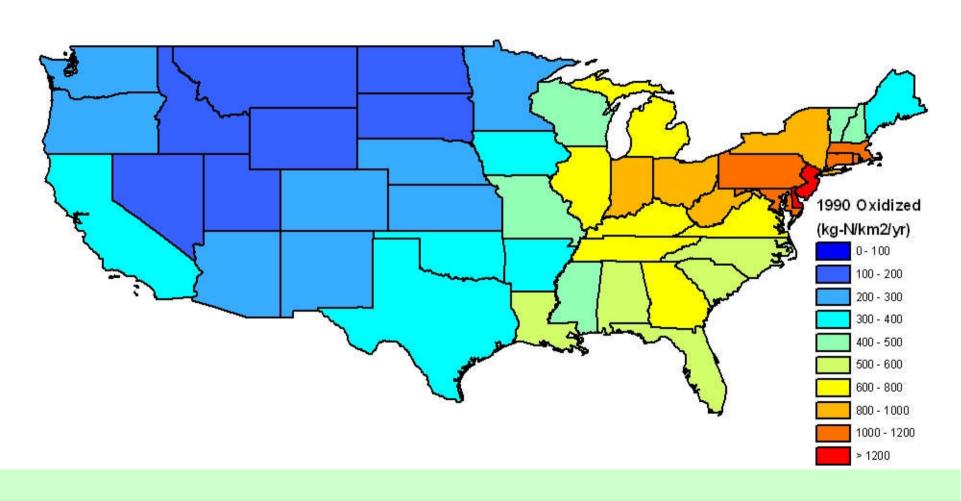
 $NO_x = Oxidized N emissions (kg N km⁻² yr⁻¹)$

Oxidized N: a=79.92; b=0.0537; c=0.0688; $r^2=0.90$

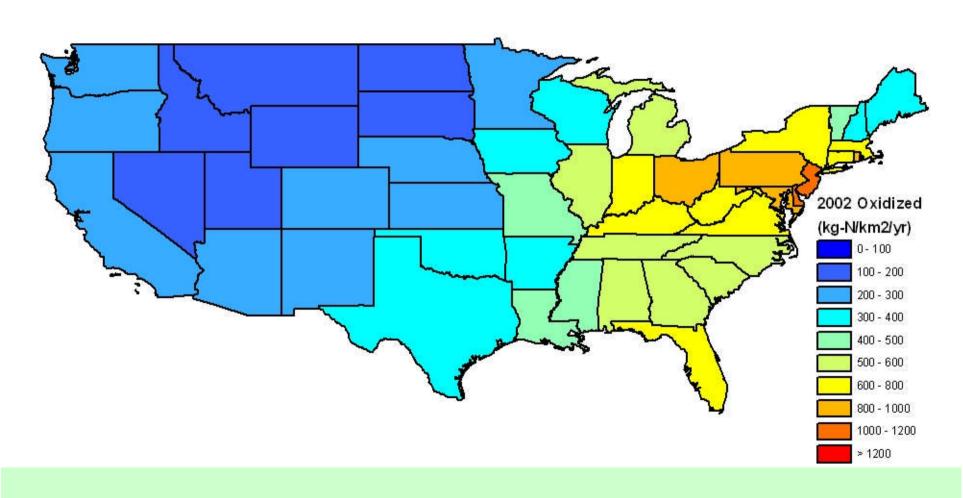
* not significantly different from zero



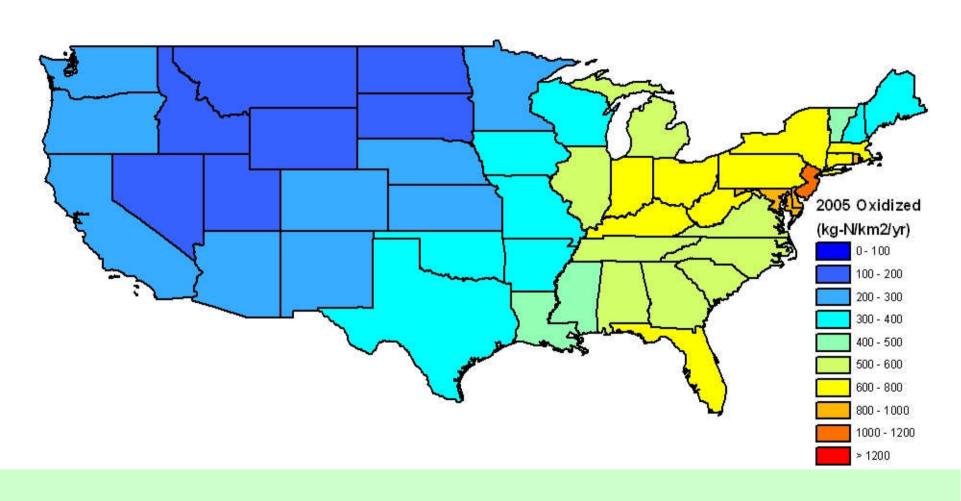
1990 predicted state level oxidized N deposition (from regression)



2002 predicted state level oxidized N deposition (from regression)



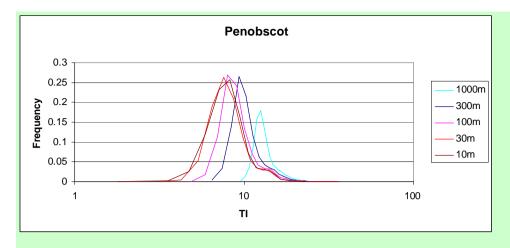
2005 predicted state level oxidized N deposition (from regression)

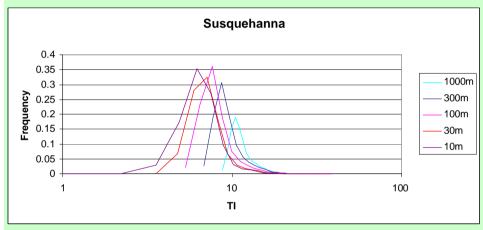


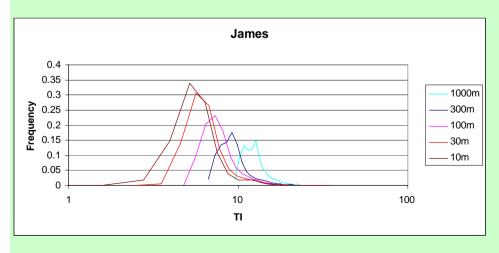
Topographic Index (Bevens and Kirkby, 1979):

$$TI_i = \ln\left(\frac{A_i}{S_i}\right)$$

Where: TI_i is the topographic index of grid cell i of the watershed, A_i is the contributing area of grid cell i of the watershed per unit contour length, s_i is the slope of grid cell i of the watershed

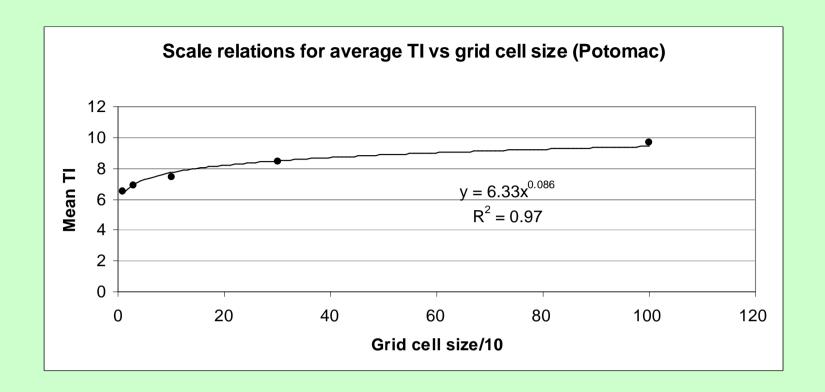




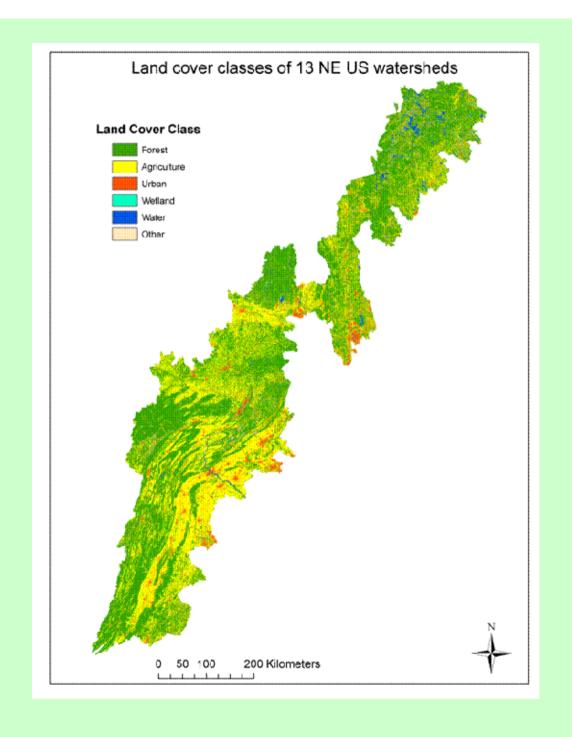


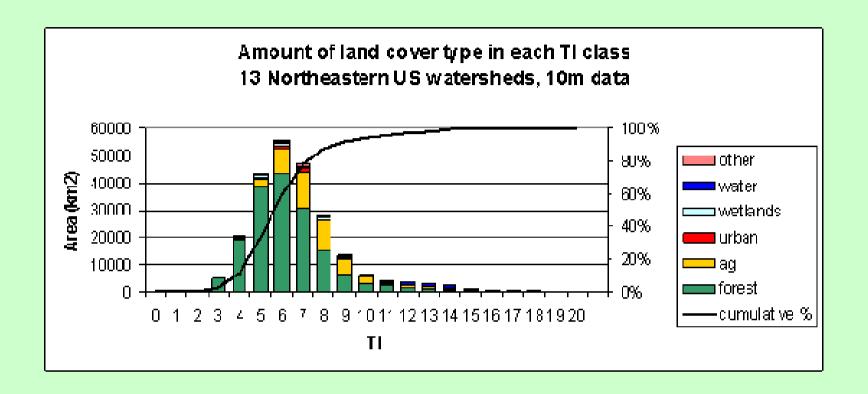
TI is scale-dependent.

The distributions at the highest resolution are lognormal and have lower mean TI (are drier) than distributions calculated at coarser scale.

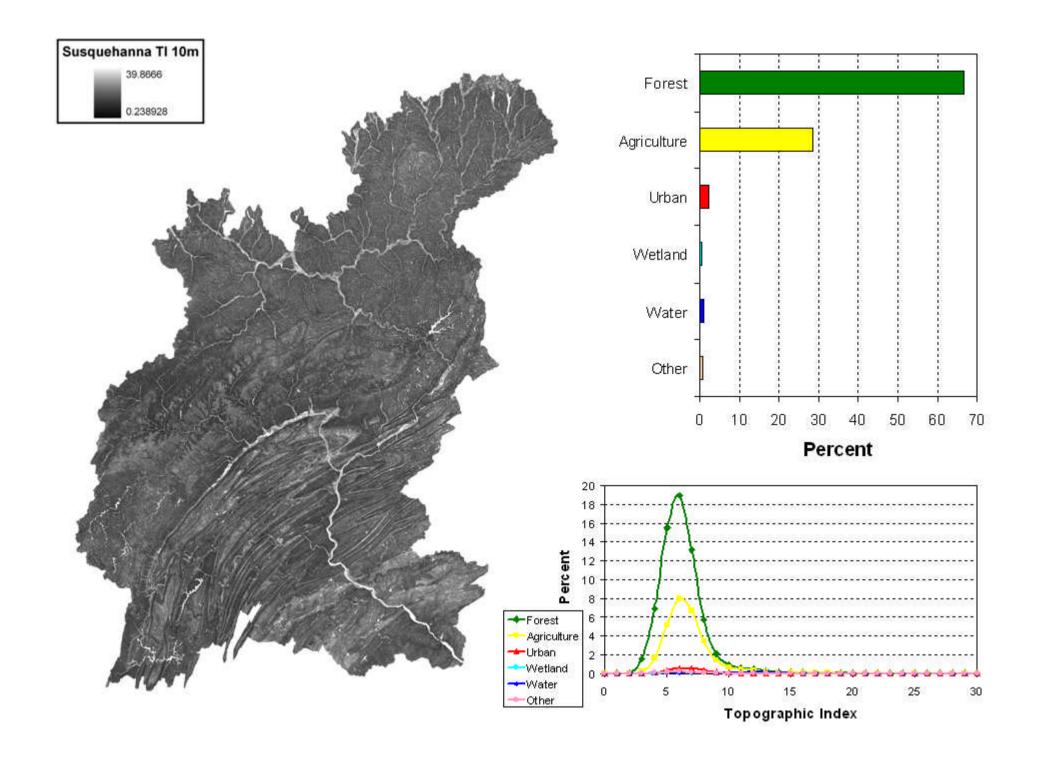


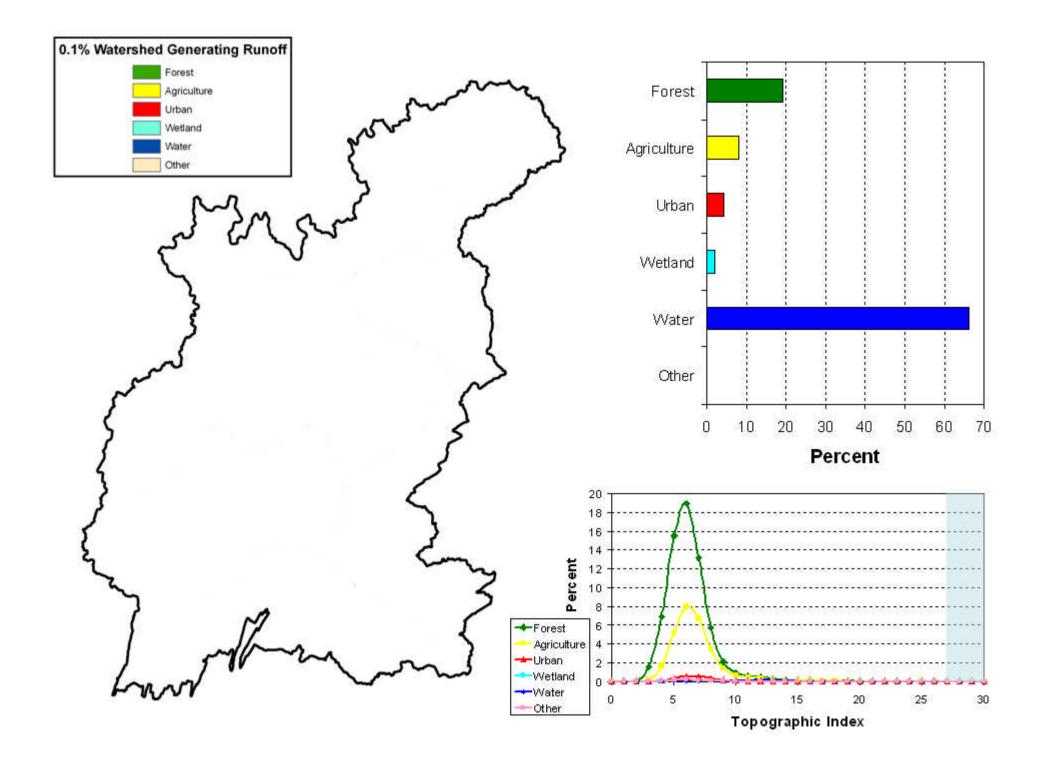
When necessary (ie in watersheds lacking fine-scale data), it may be possible to scale coarse-scale information to obtain statistics appropriate to fine-scale resolution based on scaling relations developed for these watersheds.

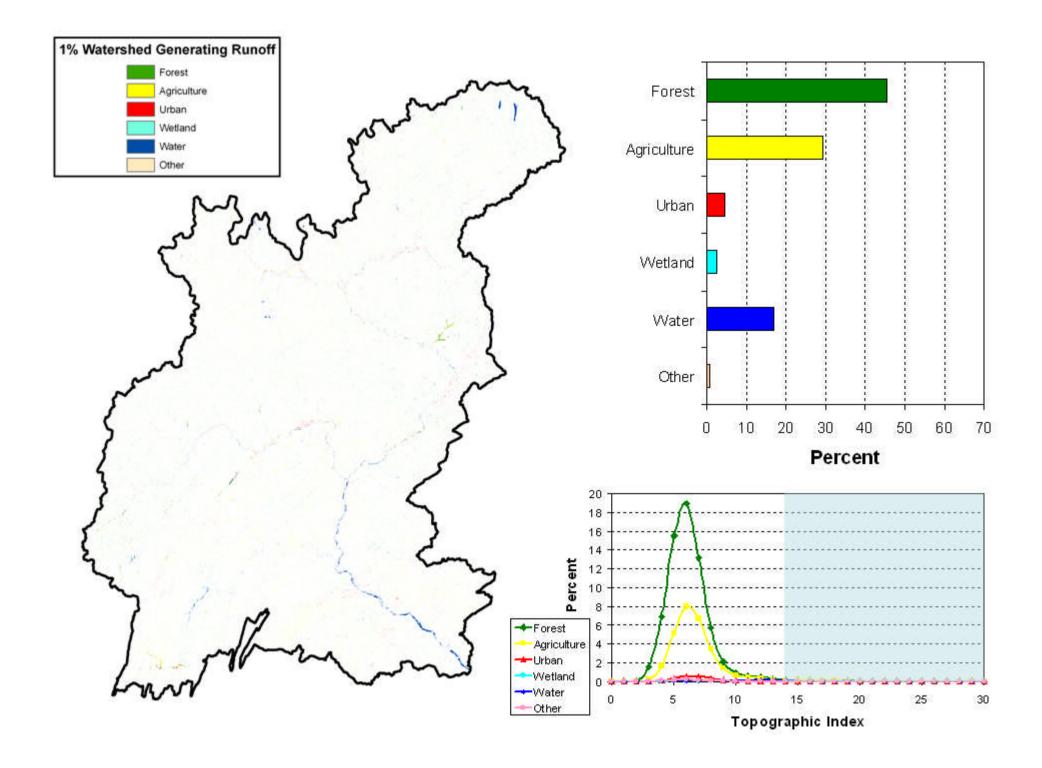


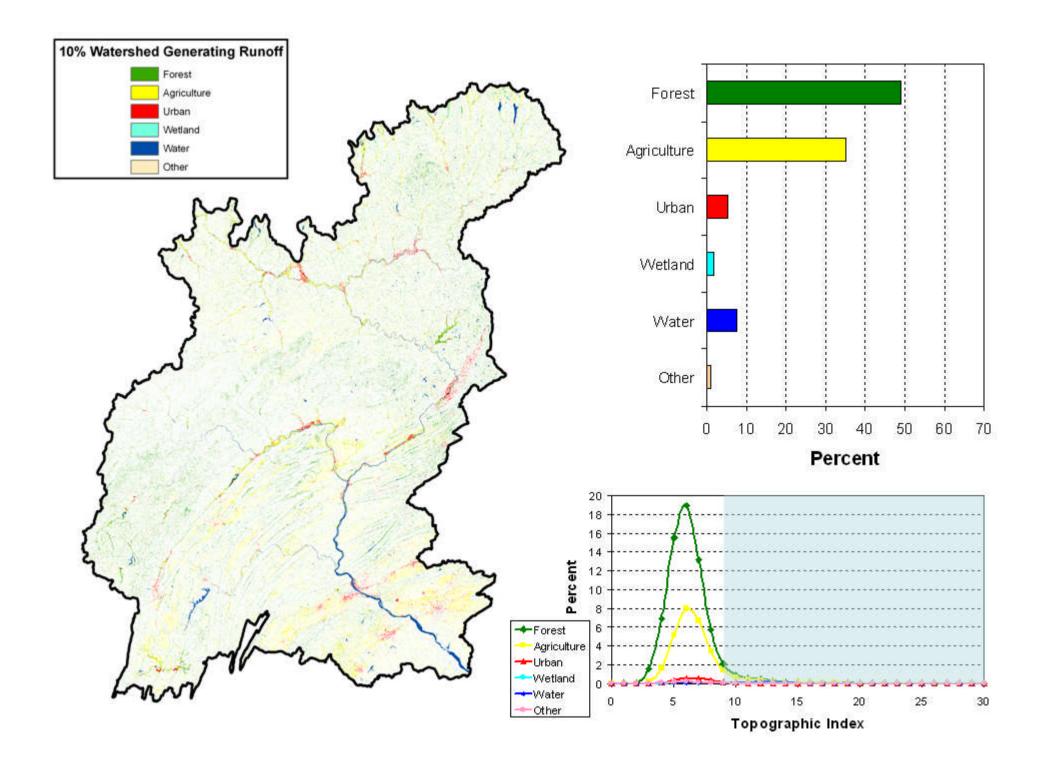


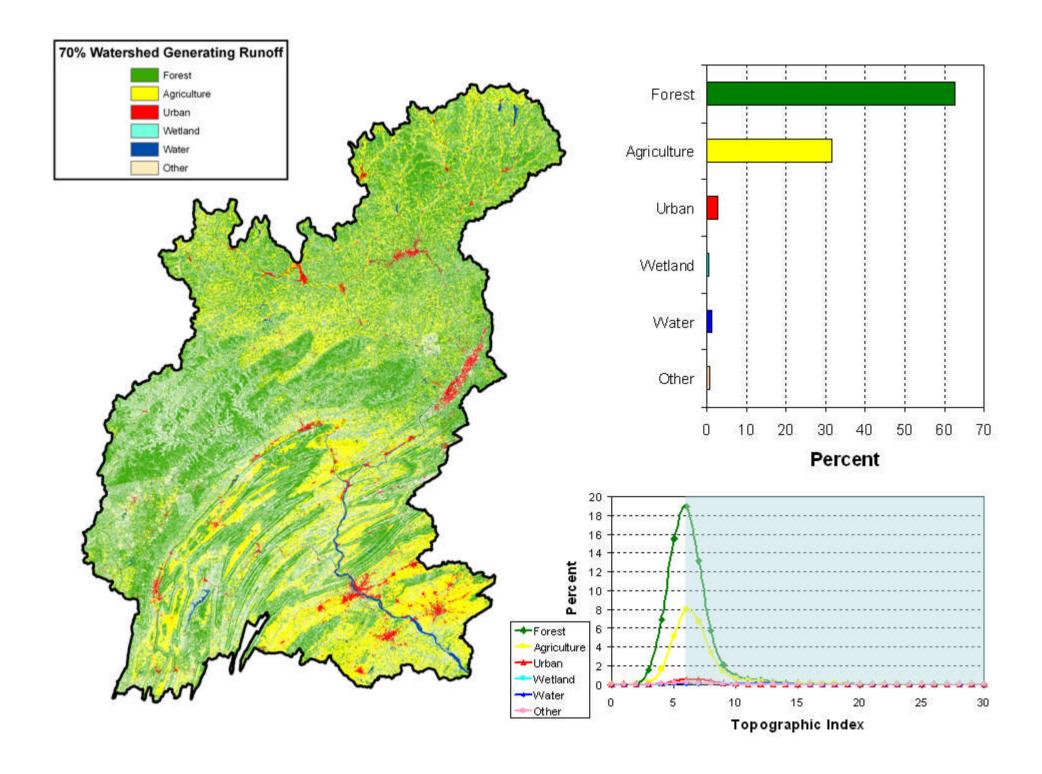
Distribution of topographic index over the 13 watersheds analyzed at 10m resolution. While a small proportion of large TI values (ie > 20) are present in the region, 99.9% of the occurrences fall between 0 and 20. High values of TI indicate wetter areas (with higher runoff potential); low values represent drier areas.











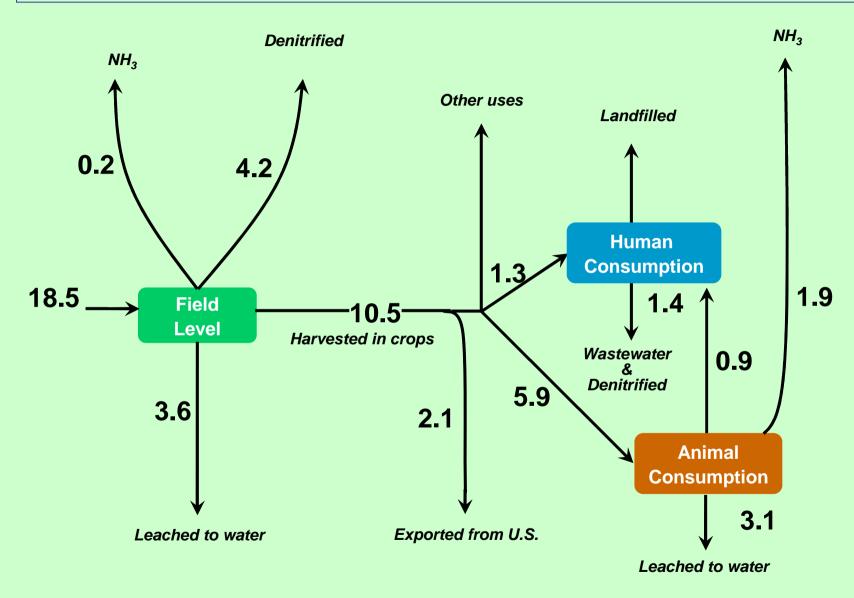




Coastal Hypoxia Research Program



Farm nitrogen balance for US (~1995; Tg per year for entire US)



(Howarth et al. 2002)